

Microwave synthesis and characterisation of tin dioxide (SnO₂) coated multi-walled carbon nanotubes

S MOTSHEKGA¹, S PILLAI¹, S RAY¹, R KRAUSE², K JALAMA²

¹National Centre for Nano-Structured Materials, PO Box 395, Pretoria, 0001, South Africa

²University of Johannesburg, PO Box 17011, Doornfontein, 2028, South Africa

Email: smotshekga@csir.co.za – www.csir.co.za

INTRODUCTION

Metal nanoparticles coated onto carbon nanotubes (CNTs) provide a way to obtain novel materials with useful properties for gas sensing and catalyst application. Metal oxides are well-known materials suitable for detecting a wide spectrum of gases with enough sensitivity. Among the materials functionalised by CNTs, intensive attention is on Tin Dioxide (SnO₂) because of its unique properties such as high optical transparency, electrical conductivity and chemical sensitivity, which made it become an attractive material for application in solar cells, field effect transistors, catalysis and gas-sensing¹⁻³. Nanostructured composite materials of carbon nanotubes and metal oxides promise superior performance over conventional approaches due to the ability to direct the selective uptake of gaseous species based on their controlled pore size and chemical properties, increased adsorptive capacity due to their increased surface area, and the effectiveness of carbon nanotubes as matrix materials for gas and vapour detection. In addition, CNTs can be functionalised, doped with catalysts and mixed with polymers accordingly to achieve the selectivity.

In this research, microwave irradiation was used in the fabrication of nanocomposites as a novel technique to disperse SnO₂ nanoparticles uniformly on multi-walled carbon nanotubes (MWCNTs) surface to composite properties. The advantage of using microwave irradiation is the rapid volumetric heating, higher reaction rate, reducing reaction time, uniform surface decoration and smaller particle size. This method of preparation was compared with the conventional technique 'wet impregnation'. Characterisation of the composites was done using techniques such as Raman, transmission electron microscopy (TEM) and scanning electron microscopy (SEM).

EXPERIMENTAL

Synthesis

MWCNTs were refluxed with 5M nitric acid (HNO₃) at 120 °C for 2 hours, filtered and then washed with distilled water until the pH of the solution was neutral. The MWCNTs were then dried at 110 degree Celsius for 12 hours. Subsequently, tin (II) chloride (SnCl₂) was dissolved in distilled water and then a certain amount of concentrated hydrochloric acid (HCl) (37%) was added. The acid treated MWCNTs were dispersed in the above solution. This mixture was sonicated for 5-10 minutes and then stirred for 60 minutes at room temperature. The precipitate was then separated from the mother liquor by centrifugation and was washed with distilled H₂O several times, and then dried at 110 degree Celsius for 12 hours. Part of the final product was calcined at 500 degree Celsius for 2 hours.

Characterisation

The morphology and microstructure of CNTs were characterised by TEM and SEM. For TEM and SEM sample preparation, the CNTs were dispersed in ethanol by ultrasonic to form a suspension. After about 3 minutes, one or two droplets were dropped onto a carbon-coated copper grid. Raman spectra were also observed for the identification degree of graphitisation of CNTs and identification of disorder in the material.

RESULTS

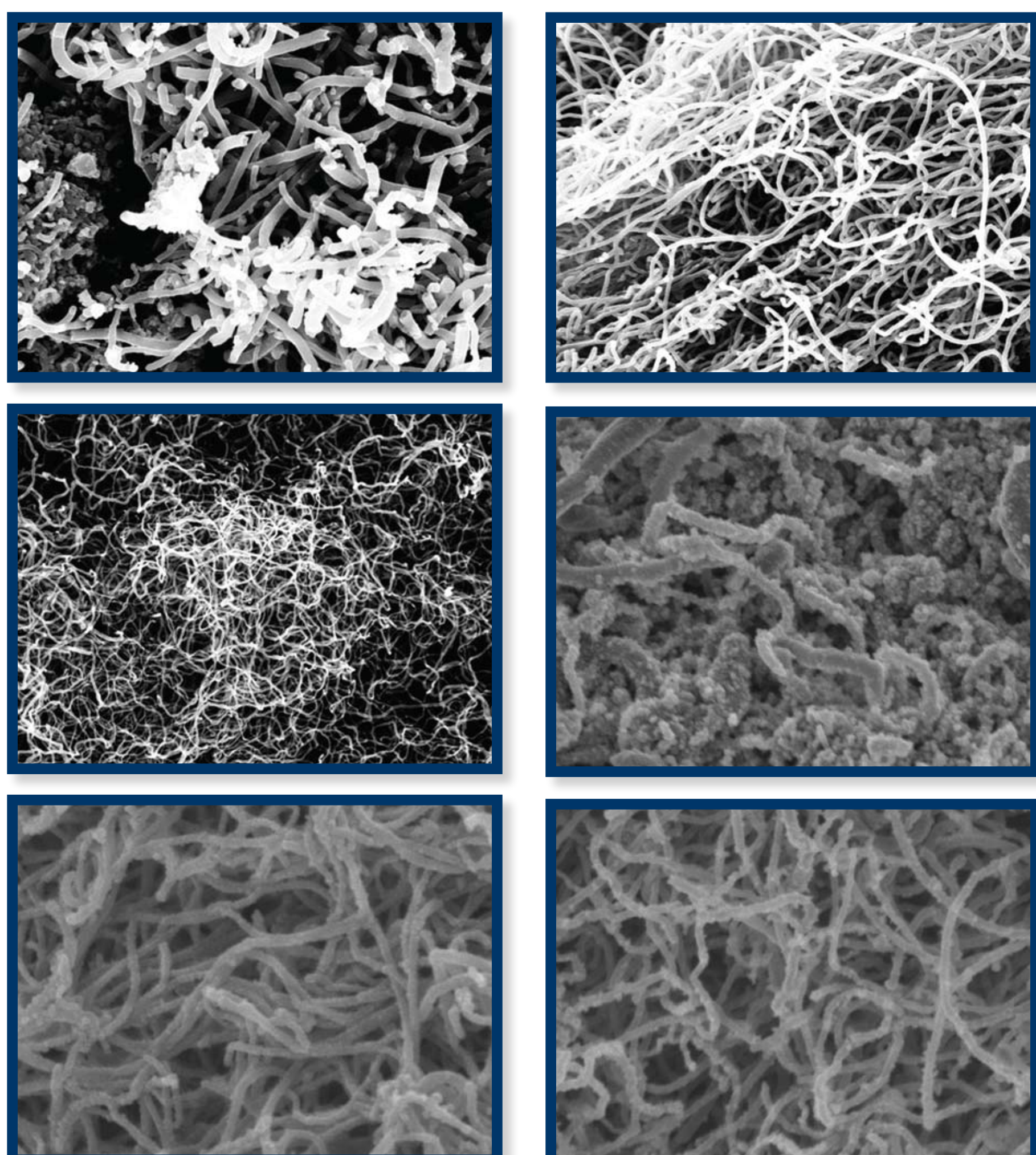


Figure 1: Scanning electron microscopy images of pure and carbon nanotubes densely coated with tin dioxide nanoparticles which are observed to be uniform around the carbon nanotubes

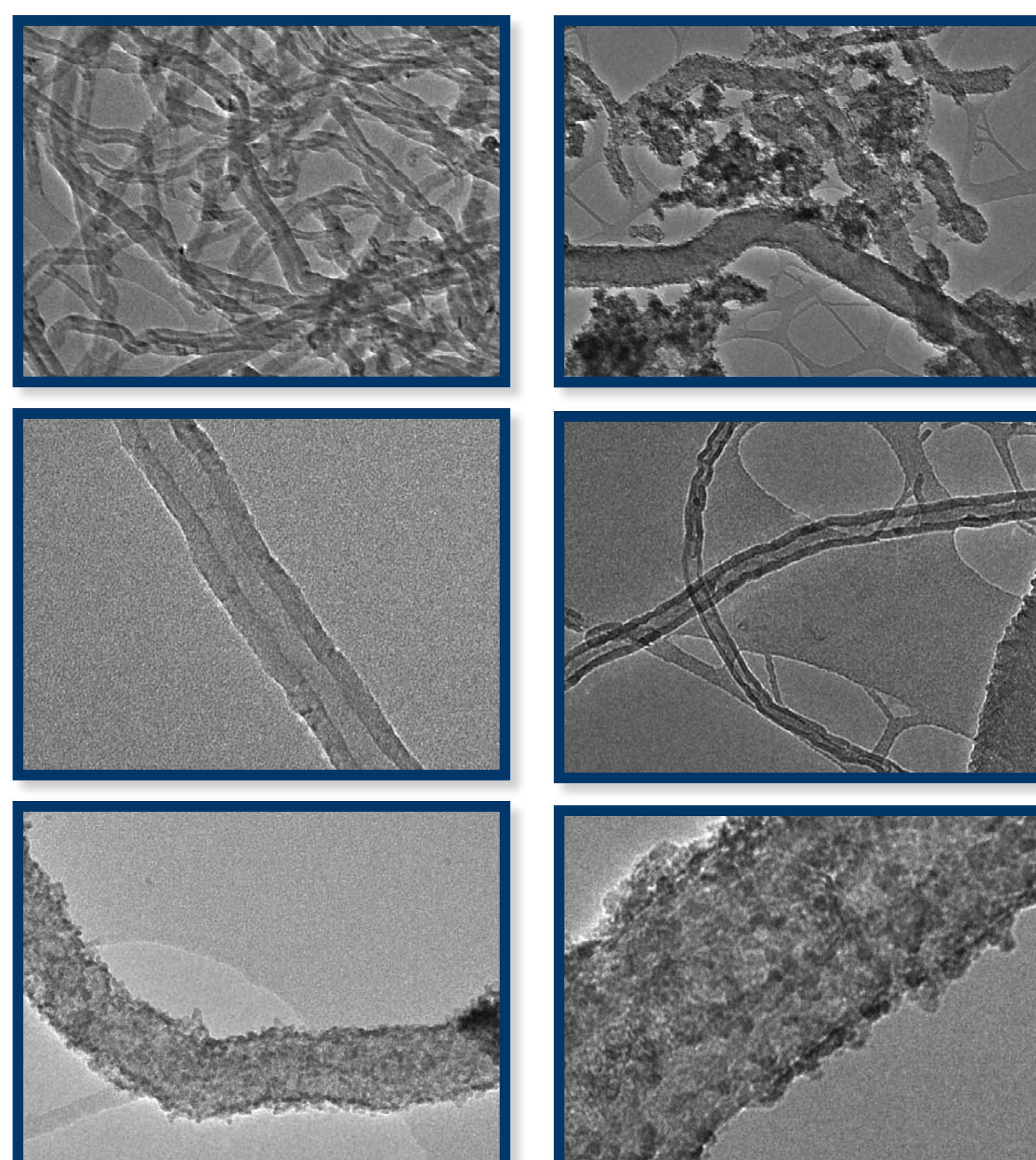


Figure 2: Transmission electron microscopy images of pure and tin dioxide coated carbon nanotubes. Tin dioxide nanoparticles deposition appears to be uniform on the surface and inside the tubes

CONCLUSION

The SnO₂ decorated MWCNTs were successfully prepared by the microwave irradiation method and compared with the wet impregnation method and the objectives of the project were met.

Developing devices through nanotechnology that detects hazardous gasses in mines and industrial processes.

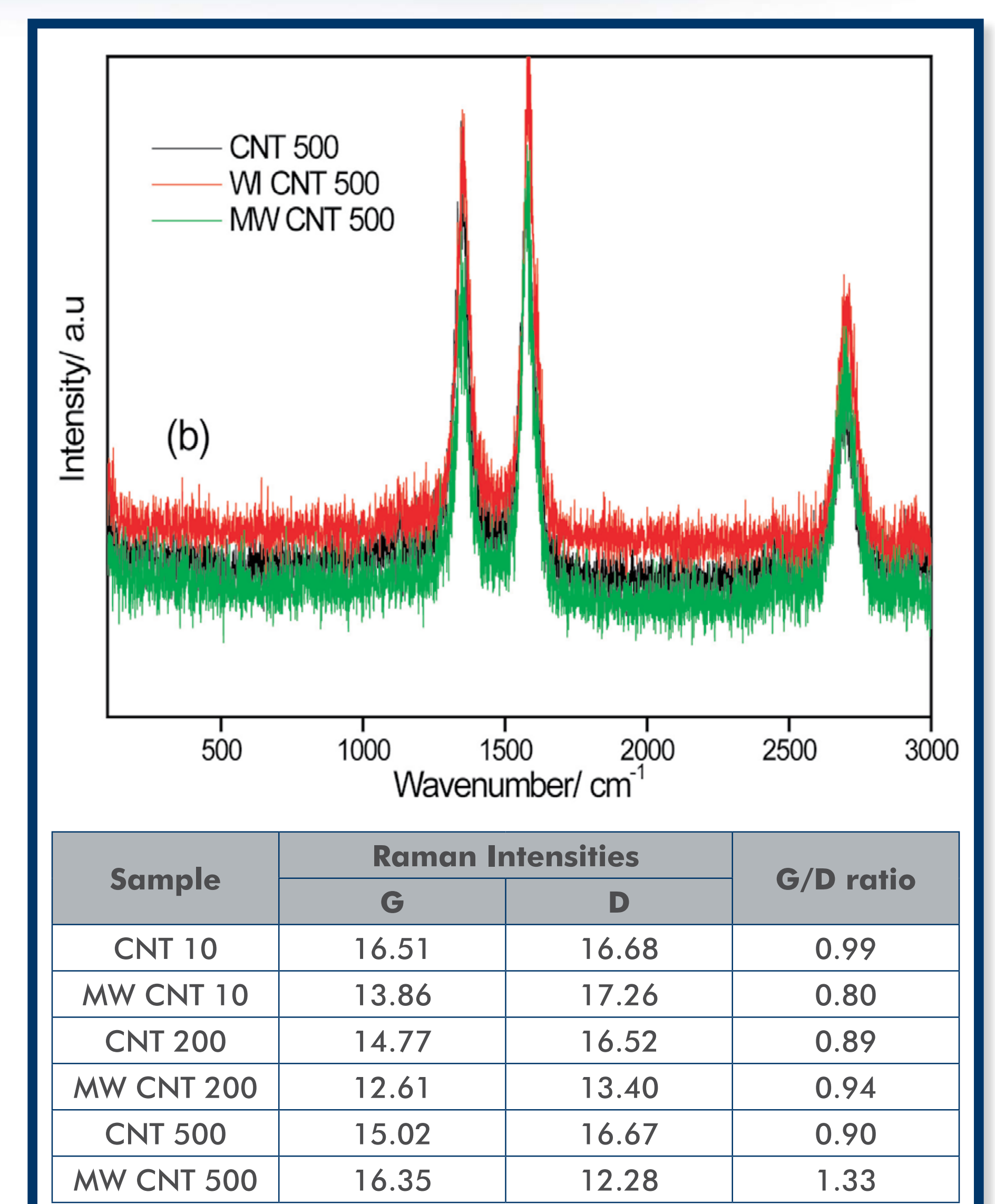


Figure 3: Raman spectra recorded for the pure carbon nanotube and SnO₂-MWCNTs composites. The decrease in G/D intensity ratio (refer to the Table) after surface modification is due to the interaction between tin dioxide nanoparticles and the surface groups of multi-walled carbon nanotubes.

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